

Name _____ Section _____

Partner _____

CHM 111 Standardization of NaOH Lab Report Form

Write the balanced equation for this titration. Circle the analyte.

As always, attach one sample of each type of calculation. Watch sig figs!

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5
Mass pure KHP (g)					
Moles pure KHP					
Volume of NaOH used (mL)					
M NaOH					

Average M _____

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5
Deviation					

Average deviation _____

Be sure to indicate if any data were omitted from the calculations and your reasoning in leaving them out.