## **CSC 335**

## Homework Set 4

Due 5:00 p.m., October 24, 2011

## Problems from Burden and Faires:

Section 4.1: #21, #26

**Section 4.2:** #1(a), #1(c), #2(a), #2(c)

**Section 4.3:** #1(c), #1(h), #3(c), #3(h), #5(c), #5(h), #7(c), #7(h)

**Section 4.4:** #1(c), #1(h), #3(c), #3(h)

**Section 4.5:** #1(c), #1(e), #6, #16

Section 4.6: #1(a), #1(e), #2(a), #2(e)

Section 4.7: #1(a), #1(d), #4(a), #4(d),

**Section 4.9:** #3(b), #3(c), #4(a)

## Problem from Dr. Pounds:

The expectation value operator is often used in quantum mechanics to return the average value of a quantum mechanical observable. In this problem we will compute the expectation value for the position of an electron trapped in a 2s orbital of the hydrogen atom. The expectation value for r is obtained by evaluating the integral:

$$\langle r \rangle = \langle \psi_{2,0,0}^* | r | \psi_{2,0,0} \rangle = \int_0^\infty \int_0^{2\pi} \int_0^\pi \psi_{2,0,0}^* r \psi_{2,0,0} r^2 \sin \theta d\theta d\phi dr$$

$$= 4\pi \int_0^\infty \psi_{2,0,0}^* r \psi_{2,0,0} r^2 dr$$

$$= 4\pi \int_0^\infty r^3 \psi_{2,0,0}^* \psi_{2,0,0} dr$$

$$= 4\pi \int_0^\infty R(r) dr$$

Where  $\psi_{2,0,0}$  is the wavefunction of the electron trapped in a 2s orbital and given by the equation:

$$\psi_{2s} = \psi_{2,0,0} = \frac{1}{4\sqrt{2\pi}} \left(\frac{Z}{a_o}\right)^{3/2} \left(2 - \frac{Z}{a_o}r\right) e^{-\left(\frac{Z}{a_o}r\right)/2}$$

Z is the atomic number and, in this case, is equal to 1. The constant  $a_o$  is called the *Bohr radius* and takes on the value of  $0.529 \times 10^{-10}$  meters. The triple integral above can be reduced to one integral because the wavefunction does not depend on the angular variables. Write a program in either JAVA, C/C++, or Fortran employing 3-, 7-, and 15-point Gauss-Laguerre quadrature<sup>1</sup> to determine the expectation value for the position of the electron in the 2s orbital of the hydrogen atom. To make things easier to interpret, it is preferable to determine the expectation value in terms of  $a_o$ . The easiest way to accomplish this is to evaluate  $R(a_o r)$  rather than R(r). If you have questions, ask Dr. Pounds.

<sup>&</sup>lt;sup>1</sup>The abcissas and weights needed for this will be placed on the class website. You will not have to generate them.